

Making our world more productive

Safety advice. Compressed gases

Safety Data Sheet

CORGON® 1252 Issue Date: 01 August 2014 Revision No: 01 Revision Date: 01-January-2022

Version: 01

1. PRODUCT AND COMPANY IDENTIFICATION

Trade Name CORGON® 12S2 UN-No UN 1956

Recommended Use Gas Metal Arc Welding

Manufacturer's Registered Office Oxygen House,

P-43 Taratala Road, Kolkata - 700088

Telephone Number (+91 33) 66021600

24 Hour Emergency Contact No: (+91) 9831851034

2. HAZARDS IDENTIFICATION

WARNING!

EMERGENCY OVERVIEW

Simple asphyxiant - this product does not contain oxygen and may cause suffocation if released in a confined area. Maintain oxygen levels above 19.5%

High pressure compressed gas Keep at temperatures below 52°C / 125°F

Appearance Colorless Physical State Compressed gas

Odor Odorless

Potential Health Effects

Principal Routes of Exposure

Inhalation.

Acute Toxicity

Inhalation

Simple asphyxiant. May cause suffocation by displacing the oxygen in the air. Exposure to an oxygen-deficient atmosphere (<19.5%) may cause dizziness, drowsiness, nausea, vomiting, excess salivation, diminished mental

alertness, loss of consciousness, and death. Exposure to atmospheres containing

8-10% or less oxygen will bring about unconsciousness without warning and so quickly that the individuals cannot

help or protect themselves. Lack of sufficient oxygen may cause serious injury or death.

Depending on concentration and duration of exposure to carbon dioxide may cause increased respiration, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptomsof overexposure

become more apparent when atmospheric oxygen is decreased to 15-17%.

Eyes None known. Contact with rapidly expanding gas near the point of release may cause severe harm.

Skin None known.

Skin Absorption Hazard No known hazard by skin absorption.

Ingestion None known.

Chronic Effects Chronic harmful effects are not known from repeated inhalation of concentrations below PEL/TLV.

Aggravated Medical Conditions Respiratory disorders.

Environmental Hazard See Section 12 for additional Ecological Information.

3. COMPOSITION/INFORMATION ON INGREDIENTS

Chemical Name	CAS-No	Volume %	Chemical Formula
Argon	7440-37-1	86	Аг
Carbon dioxide	124-38-9	12	CO ₂
Oxygen	7782-44-7	2	0,

4. FIRST AID MEASURES

Eye Contact None under normal use. Get medical attention if symptoms occur.

Skin Contact None under normal use. Get medical attention if symptoms occur.

Inhalation PROMPT MEDICAL ATTENTION IS MANDATORY IN ALL CASES OF INHALATION OVEREXPOSURE. RESCUE PERSONNEL

SHOULD BE EQUIPPED WITH SELF-CONTAINED BREATHING APPARATUS (SCBA).

Conscious inhalation victims should be assisted to an uncontaminated area and inhale fresh air. If breathing is difficult, Administer oxygen under medical supervision / trained personnel supervision. Unconscious persons should be moved to an uncontaminated area and as necessary, given artificial resuscitation and supplemental

oxygen. Treatment should be symptomatic and supportive.

Ingestion

None under normal use. Get medical attention if symptoms occur.

5. FIRE-FIGHTING MEASURES

Flammable Properties Not flammable. Does not support combustion.

Suitable Extinguishing MediaUse extinguishing agent suitable for type of surrounding fire.

Explosion Data

Sensitivity to Mechanical Impact None

Sensitivity to Static Discharge None

Specific Hazards Arising from the

Chemical

Cylinders may rupture under extreme heat. Continue to cool fire-exposed cylinders until flames are extinguished. Damaged cylinders should be handled only by specialists.

Protective Equipment and Precautions for Firefighters

6. ACCIDENTAL RELEASE MEASURES

Personal Precautions Ensure adequate ventilation. Evacuate personnel to safe areas. Use personal protective equipment.

Monitor oxygen level.

Environmental Precautions Prevent the spreading of vapors through sewers, ventilation systems, and confined areas.

Methods for ContainmentStop the flow of gas or remove the cylinder to an outdoor location if this can be done without risk. If a leak is in the

container or container valve, contact the appropriate emergency telephone number in Section 1 or callyour

closest Linde location.

Methods for Cleaning Up Return cylinder to Linde India Ltd.

OtherInformation Ventilate the area.

7. HANDLING AND STORAGE

Handling Use only in ventilated areas. Never attempt to lift a cylinder by its valve protection cap.

Protect cylinders from physical damage; do not drag, roll, slide or drop. When moving cylinders, even for a short distance, use a trolley designed to transport cylinders. Use equipment rated for cylinder pressure. Use a backflow preventive device in the piping. Never insert an object (e.g., wrench, screwdriver, pry bar, etc.) into valve cap openings. Doing so may damage the valve, causing a leak to occur.

Close valve after each use and when empty. If a user experiences any difficulty operating the cylinder valve discontinue use and contact the supplier.

Never put cylinders into trunks of cars or unventilated areas of passenger vehicles. Never attempt to re-fill a compressed gas cylinder without the owner's written consent. Never strike an arc on a compressedgas cylinder or make a cylinder a part of an electrical circuit.

For additional recommendations consult rule number 20 of the Gas Cylinders, Rules, 2016.

Protect from physical damage. Cylinders should be stored upright with a valve protection cap in place and firmly

secured to prevent falling. Store in a cool, dry, well-ventilated area of non-combustible construction away from

high traffic areas and emergency exits. Keep at temperatures below 52°C / 125°F .

Full and empty cylinders should be segregated. Use a "First-In-First-Out" (FIFO) inventory system to prevent full cylinders from being stored for excessive periods of time. Always store and handle compressed gas cylinders in accordance with rule number 21 of the Gas Cylinders, Rules, 2016.

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Storage

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

Exposure Guidelines

Chemical Name	ACGIH TLV	OSHA PEL	NIOSHIDLH
Carbon dioxide 124-38-9	STEL = 30000 ppm TWA: 5000 ppm	TWA: 5000 ppm TWA: 9000 mg/m3 (vacated) TWA: 10000 ppm (vacated) TWA: 18000 mg/m³(vacated) STEL: 30000 ppm	IDLH: 40000 ppm TWA: 5000 ppm TWA: 9000 mg/m ³ STEL: 54000 mg/m ³ STEL: 30000 ppm

Immediately Dangerous to Life or Health.

Other Exposure Guidelines Vacated limits revoked by the Court of Appeals decision in AFL-CIO v. OSHA, 965 F.2d 962

(11th Cir., 1992).

Engineering Measures Local exhaust ventilation to prevent accumulation of high concentrations and maintain air-oxygen levelsat or

above 19.5%.

Ventilation Use ventilation adequate to keep exposures below recommended exposure limits.

Personal Protective Equipment

Eye/Face Protection Wear protective eyewear (safety glasses).

Skin and Body Protection Work gloves and safety shoes are recommended when handling cylinders.

Respiratory Protection

General UseNo respiratory equipment is needed if workplace oxygen levels are kept above 19.5%.

Emergency UseUse positive pressure airline respirator with escape cylinder or self-contained breathing apparatus foroxygen-

deficient atmospheres (<19.5%).

Hygiene Measures Handle in accordance with good industrial hygiene and safety practices.

9. PHYSICAL AND CHEMICAL PROPERTIES

Product Information

AppearanceColorless.OdorOdorless.Odor ThresholdNo information available.Physical StateCompressed gasFlash PointNot applicable.Autoignition TemperatureNo information available.

Flammability Limits in Air

UpperNot applicableLowerNot applicable

The following information is for the NON-INERT components of this mixture

Chemical Name	Boiling Point	Melting Point	Molecular Weight	Evaporation Rate	WaterSolubility	Vapor Pressure	Vapor Density (Air=1)	Gas Density Kg/m3@20°C
Oxygen	-183°C	-219°C	32.00	-	Slightly soluble	Above critical temperature	1.11	1.33
Carbon dioxide	56°C	-56°C	44.00	-	0.145 g/ml @ 25°C	838 psig (5778 kPa) @ 21.1°C	1.52	1.84

The following information is for the INERT components that may be part of this mixture:

Chemical Name	Boiling Point	Melting Point	Molecular	Evaporation Rate	Water Solubility	Vapor Pressure	Vapor Density (Air=1)	Gas Density Kg/m3@20°C
Argon	-185.9°C	-189.4°C	39.94	-	0.056 (vol/vol @ 0°C and 1 atm)	Above critical temperature	1.38	1.65

10. STABILITY AND REACTIVITY

Stability Stable.

Incompatible Products Carbon dioxide is incompatible with certain reactive metals, hydrides etc and may ignite. Passing carbon dioxide

over a mixture of sodium peroxide and aluminum or magnesium may explode.

Conditions to Avoid Due to the presence of Carbon dioxide, Carbonic acid is formed in the presence of moisture.

Hazardous Decomposition ProductsNone known based on information supplied.

Hazardous PolymerizationHazardous polymerization does not occur.

11. TOXICOLOGICAL INFORMATION

Acute Toxicity Product

Information

LD50 Oral: No information available.

LD50 Dermal: No information available.

LC50 Inhalation: No information available.

Repeated Dose Toxicity No information available.

Component Information No information available.

Chemical Name	LD50 Oral	LD50 Dermal	LC50 Inhalation
Carbon dioxide			470000 ppm (Rat)

Toxicity

Chronic Toxicity Chronic harmful effects are not known from repeated inhalation of concentrations below PEL/TLV.

Carcinogenicity Contains no ingredient listed as a carcinogen.

Irritation No information available.

Sensitization No information available.

Reproductive Toxicity No information available.

Developmental ToxicityOxygen deficiency during pregnancy has produced developmental abnormalities in humans and experimental

animals.

Synergistic Materials None known.

Target Organ Effects Central vascular system (CVS), Respiratory system.

12. ECOLOGICAL INFORMATION

Ecotoxicity

The environmental impact of this product has not been fully investigated.

Ozone depletion potential; ODP; (R-11 = 1): Does not contain ozone depleting chemical.

13. DISPOSAL CONSIDERATIONS

Waste Disposal Methods Do not attempt to dispose of residual waste or unused quantities. Return in the shipping container PROPERLY

LABELED WITH ANY VALVE OUTLET PLUGS OR CAPS SECURED AND VALVE PROTECTION CAP IN

PLACE to Linde India Ltd for proper disposal.

Contaminated Packaging Do not re-use empty containers.

14. TRANSPORT INFORMATION

Proper shipping name Compressed gas, n.o.s.

Hazard Class 2.2 Subsidiary Class None UN-Number UN1956

Description UN1956, Compressed gas, n.o.s.(Argon, Carbon Dioxide, Oxygen), 2.2

15. Regulatory Information

Labeling of cylindersLabel 2.2: non-flammable non- toxic gas.Risk phrasesRAs Asphyxiate in high concentrations.Safety phrasesS9 Keep container in a well-ventilated place.

S23 Do not breathe gas.

16. OtherInformation



General Ensure all national/local regulations are observed.

The hazard of asphyxiation is often overlooked and must be stressed during operator training.

Document Information In preparing this document help has been taken from MSDS for Linde (US)

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End of Safety Data Sheet